Cambridge Assessment International Education

Cambridge International AS & A Level

PHYSICS **9702/43**

Paper 4 A Level Structured Questions **October/November 2020**

MARK SCHEME

Maximum Mark: 100

Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the October/November 2020 series for most Cambridge IGCSE™, Cambridge International A and AS Level and Cambridge Pre-U components, and some Cambridge O Level components.

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Generic Marking Principles

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptors for a question. Each question paper and mark scheme will also comply with these marking principles.

GENERIC MARKING PRINCIPLE 1:

Marks must be awarded in line with:

- the specific content of the mark scheme or the generic level descriptors for the question
- the specific skills defined in the mark scheme or in the generic level descriptors for the question
- the standard of response required by a candidate as exemplified by the standardisation scripts.

GENERIC MARKING PRINCIPLE 2:

Marks awarded are always **whole marks** (not half marks, or other fractions).

GENERIC MARKING PRINCIPLE 3:

Marks must be awarded **positively**:

- marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate
- marks are awarded when candidates clearly demonstrate what they know and can do
- marks are not deducted for errors
- marks are not deducted for omissions
- answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.

GENERIC MARKING PRINCIPLE 4:

Rules must be applied consistently, e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

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GENERIC MARKING PRINCIPLE 5: Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).
GENERIC MARKING PRINCIPLE 6: Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

Science-Specific Marking Principles

1	Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.
2	The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.
3	Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).
4	The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.
5	<u>'List rule' guidance</u> For questions that require ***n*** responses (e.g. State **two** reasons …): • The response should be read as continuous prose, even when numbered answer spaces are provided. • Any response marked *ignore* in the mark scheme should not count towards ***n***. • Incorrect responses should not be awarded credit but will still count towards ***n***. • Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response. • Non-contradictory responses after the first ***n*** responses may be ignored even if they include incorrect science.

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6 Calculation specific guidance

Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.

For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.

For answers given in standard form (e.g. $a \times 10^n$) in which the convention of restricting the value of the coefficient (a) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.

Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.

7 Guidance for chemical equations

Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.

State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

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Abbreviations

/	Alternative and acceptable answers for the same marking point.
( )	Bracketed content indicates words which do not need to be explicitly seen to gain credit but which indicate the **context** for an answer. The context does not need to be seen but if a context is given that is incorrect then the mark should not be awarded.
___	Underlined content must be present in answer to award the mark. This means either the exact word or another word that has the same technical meaning.

Mark categories

B marks	These are independent marks, which do not depend on other marks. For a **B** mark to be awarded, the point to which it refers must be seen specifically in the candidate's answer.
M marks	These are method marks upon which **A** marks later depend. For an **M** mark to be awarded, the point to which it refers must be seen specifically in the candidate's answer. If a candidate is not awarded an **M** mark, then the later **A** mark cannot be awarded either.
C marks	These are compensatory marks which can be awarded even if the points to which they refer are not written down by the candidate, providing subsequent working gives evidence that they must have known them. For example, if an equation carries a **C** mark and the candidate does not write down the actual equation but does correct working which shows the candidate knew the equation, then the **C** mark is awarded. If a correct answer is given to a numerical question, all of the preceding **C** marks are awarded automatically. It is only necessary to consider each of the **C** marks in turn when the numerical answer is not correct.
A marks	These are answer marks. They may depend on an **M** mark or allow a **C** mark to be awarded by implication.

Annotations

✓	Indicates the point at which a mark has been awarded.
X	Indicates an incorrect answer or a point at which a decision is made not to award a mark.
XP	Indicates a physically incorrect equation ('incorrect physics'). No credit is given for substitution, or subsequent arithmetic, in a physically incorrect equation.

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ECF	Indicates 'error carried forward'. Answers to later numerical questions can always be awarded up to full credit provided they are consistent with earlier incorrect answers. Within a section of a numerical question, ECF can be given after AE, TE and POT errors, but **not** after XP.
AE	Indicates an arithmetic error. Do not allow the mark where the error occurs. Then follow through the working/calculation giving full subsequent ECF if there are no further errors.
POT	Indicates a power of ten error. Do not allow the mark where the error occurs. Then follow through the working/calculation giving full subsequent ECF if there are no further errors.
TE	Indicates incorrect transcription of the correct data from the question, a graph, data sheet or a previous answer. For example, the value of 1.6×10^{-19} has been written down as 6.1×10^{-19} or 1.6×10^{19}. Do not allow the mark where the error occurs. Then follow through the working/calculation giving full subsequent ECF if there are no further errors.
SF	Indicates that the correct answer is seen in the working but the final answer is incorrect as it is expressed to too few significant figures.
BOD	Indicates that a mark is awarded where the candidate provides an answer that is not totally satisfactory, but the examiner feels that sufficient work has been done ('benefit of doubt').
CON	Indicates that a response is contradictory.
I	Indicates parts of a response that have been seen but disregarded as irrelevant.
M0	Indicates where an A category mark has not been awarded due to the M category mark upon which it depends not having previously been awarded.
^	Indicates where more is needed for a mark to be awarded (what is written is not wrong, but not enough). May also be used to annotate a response space that has been left completely blank.
SEEN	Indicates that a page has been seen.

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Question	Answer	Marks
1(a)(i)	region (of space)	**B1**
	where a particle experiences a force	**B1**
1(a)(ii)	force per unit mass	**B1**
1(b)	$g = GM / R^2$	**C1**
	$= (6.67 \times 10^{-11} \times 6.42 \times 10^{23}) / (3.39 \times 10^{6})^2$	**C1**
	$= 3.73\ \text{N kg}^{-1}$	**A1**

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Question	Answer	Marks
1(c)	$0.99 \times 3.73 = (6.67 \times 10^{-11} \times 6.42 \times 10^{23}) / r^2$	**C1**
	$r = 3.41 \times 10^6$ (m)	**C1**
	height = $(r - R)$ $= 2 \times 10^4$ m	**A1**
	or	
	$0.99 \times 3.73 = (6.67 \times 10^{-11} \times 6.42 \times 10^{23}) / (R + h)^2$ $(R + h)^2 = 1.1596 \times 10^{13}$	**(C1)**
	$R + h = 3.41 \times 10^6$ (m)	**(C1)**
	$h = 2 \times 10^4$ m	**(A1)**
	or	
	$0.99 = (3.39 \times 10^6)^2 / r^2$	**(C1)**
	$r = 3.41 \times 10^6$ (m)	**(C1)**
	height = 2×10^4 m	**(A1)**

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<table>
<tr><th>Question</th><th>Answer</th><th>Marks</th></tr>
<tr><td rowspan="2">2(a)</td><td>+q: thermal energy transfer to system</td><td>B1</td></tr>
<tr><td>+w: work done on system</td><td>B1</td></tr>
<tr><td>2(b)(i)</td><td>(W =) $2.6 \times 10^5 \times (3.8 - 2.3) \times 10^{-3}$ = 390 J</td><td>A1</td></tr>
<tr><td rowspan="2">2(b)(ii)</td><td>no (total) change (in internal energy)</td><td>B1</td></tr>
<tr><td>gas returns to its original temperature</td><td>B1</td></tr>
<tr><td rowspan="4">2(c)</td><td>A to B row all correct (1370, – 390, 980)</td><td>B1</td></tr>
<tr><td>B to C row all correct (0, 550, 550)</td><td>B1</td></tr>
<tr><td>C to A row: ΔU adds to the other two ΔU values to give zero</td><td>B1</td></tr>
<tr><td>C to A row: w = 0 and q adds to w to give ΔU value

complete correct answer:

<table>
<tr><th>change</th><th>q / J</th><th>w / J</th><th>ΔU / J</th></tr>
<tr><td>A to B
B to C
C to A</td><td>(+)1370
0
–1530</td><td>–390
(+)550
0</td><td>(+)980
(+)550
–1530</td></tr>
</table>
</td><td>B1</td></tr>
</table>

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Question	Answer	Marks
3(a)	acceleration (directly) proportional to displacement	**B1**
	acceleration in opposite direction to displacement **or** acceleration (directed) towards equilibrium position	**B1**
3(b)	$v = \omega(x_0^2 - x^2)^{½}$ **and** $\omega = 2\pi f$ **or** $v_0 = x_0\omega$ **and** $\omega = 2\pi f$	**C1**
	substitution of any correct point from graph, e.g. for $x = 0$: $0.25 = 2\pi f \times 8.8 \times 10^{-2}$	**C1**
	$f = 0.45$ Hz	**A1**
3(c)	$1 / 0.45 = 2\pi \times (L / 9.81)^{½}$	**C1**
	$L = 1.2$ m	**A1**
3(d)	ellipse about the origin with same intercepts on x-axis	**B1**
	ellipse about the origin crossing v-axis inside original loop	**B1**

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Question	Answer	Marks
4(a)	quartz crystal	**B1**
	alternating p.d. across crystal causes it to vibrate	**B1**
	resonance occurs when frequency of p.d. matches natural frequency of crystal	**B1**
	natural frequency of crystal is in ultrasound range	**B1**
4(b)	$I = I_0 e^{-\mu x}$	**C1**
	$I / I_0 = e^{-1.2 \times 3.5}$ $= 0.015$	**C1**
	ratio / dB = –10 lg (1 / 0.015) **or** 10 lg (0.015)	**C1**
	= –18 dB	**A1**

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Question	Answer	Marks
5(a)	work done per unit charge	**B1**
	(work done on charge) moving positive charge from infinity	**B1**
5(b)(i)	$(2.0 \times 10^{-9}) / 4\pi\varepsilon_0(4.0 \times 10^{-2}) + Q / 4\pi\varepsilon_0(8.0 \times 10^{-2}) = 0$	**C1**
	$Q = 4.0 \times 10^{-9}$ C	**A1**
	Q given with negative sign	**B1**
5(b)(ii)	change = 1200 V	**A1**
5(c)	$½mv^2 = qV$	**C1**
	$½ \times 4 \times 1.66 \times 10^{-27} \times v^2 = 2 \times 1.60 \times 10^{-19} \times 1200$	**C1**
	$v = 3.4 \times 10^5$ m s^{-1}	**A1**

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Question	Answer	Marks
6(a)(i)	charge per unit potential (difference)	**M1**
	charge on one plate and potential difference across the plates	**A1**
6(a)(ii)	any three points from: • smoothing • timing/(time) delay • tuning • oscillator • blocking d.c. • surge protection • temporary power supply	**B3**
6(b)	(capacitors in series have combined capacitance =) 8 μF	**C1**
	capacitance = 8 + 24 = 32 μF	**A1**

Question	Answer	Marks
7(a)	two resistors connected in series between earth and positive of battery and no extra connections	**B1**
	one resistor and thermistor connected in series between earth and positive of battery and no extra connections	**B1**
	midpoints of the two potential dividers connected, one each, to the op-amp input terminals	**B1**
	thermistor in correct place in potential divider circuit (either the upper part of the potential divider leading to V^+ or the lower part of the potential divider leading to V^-)	**B1**
7(b)(i)	value greater than 1000 Ω	**A1**
7(b)(ii)	non-zero value less than 1000 Ω	**A1**

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Question	Answer	Marks
8(a)(i)	downwards	**B1**
8(a)(ii)	PQRS and JKLM	**B1**
8(b)	(as charge separates) an electric field is created (between opposite faces)	**B1**
	(maximum value is reached when) electric force (on electron) is equal and opposite to magnetic force (on electron)	**B1**
8(c)	$V_H = BI / ntq$ $= (4.6 \times 10^{-3} \times 6.3 \times 10^{-4}) / (1.3 \times 10^{29} \times 0.10 \times 10^{-3} \times 1.60 \times 10^{-19})$	**C1**
	$= 1.4 \times 10^{-12}$ V	**A1**
8(d)	semiconductors have a (much) smaller value for n	**B1**
	V_H for semiconductors is (much) larger so more easily measured	**B1**

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Question	Answer	Marks
9(a)	flux density × area	**M1**
	where flux is normal to area	**A1**
	or	
	flux density × area × sin θ	**(M1)**
	where θ is angle between flux direction and (plane of) area	**(A1)**
9(b)(i)	(alternating) current creates changing (magnetic) flux	**B1**
	core links (magnetic) flux with secondary coil	**B1**
	changing flux (in secondary) causes induced e.m.f.	**B1**
9(b)(ii)	rate of change of flux is not constant	**B1**
	(induced) e.m.f. is proportional to rate of change of flux	**B1**
9(c)	reduces induced currents in core	**B1**
	hence reduces energy losses (in core)	**B1**

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Question	Answer	Marks
10(a)	X-rays are used	**B1**
	section (of object) is scanned	**B1**
	scans/images taken at many angles/directions **or** images of each section are 2-dimensional	**B1**
	(images of (many)) sections are combined	**B1**
	(to give) 3-dimensional image of (whole) structure	**B1**
10(b)	K = 6 L = 7 M = 2 N = 9 3 marks: all four correct 2 marks: three correct and one incorrect **or** all correct with two numbers transposed 1 mark: two correct and two incorrect	**B3**

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Question	Answer	Marks
11(a)(i)	quantum of energy	**M1**
	of electromagnetic radiation	**A1**
11(a)(ii)	arrow (on Fig. 11.1) pointing upwards and to the right	**B1**
11(b)(i)	$\lambda = h / p$	**C1**
	$p = (6.63 \times 10^{-34}) / (544 \times 10^{-9})$ $= 1.22 \times 10^{-27}$ N s	**A1**
11(b)(ii)	energy $= hc / \lambda$	**C1**
	$= 6.63 \times 10^{-34} \times 3.00 \times 10^{8} \times (540^{-1} - 544^{-1}) \times 10^{9}$ $= 2.7 \times 10^{-21}$ J	**A1**
11(c)	(smaller wavelength corresponds to) greater photon energy	**B1**
	any one point from: • (deflected) photon loses energy (so not possible) • (deflected) photon would need to gain energy (so not possible) • electron would need to lose energy (so not possible) • initially electron energy is zero (so not possible)	**B1**

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Question	Answer	Marks
12(a)(i)	unstable nucleus	**B1**
	emits ionising radiation **or** decays spontaneously	**B1**
12(a)(ii)	probability of decay (of a nucleus)	**M1**
	per unit time	**A1**
12(b)	$A = \lambda N$	**C1**
	$560 = 9.9 \times 10^{-7} \times N$ $N = 5.7 \times 10^{8}$	**A1**
12(c)	$A = A_0 e^{-\lambda t}$ $170 = 560 \exp(-9.9 \times 10^{-7} \times t)$	**C1**
	$t = 1.2 \times 10^{6}$ s	**C1**
	= 14 days	**A1**

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